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Solar g Flat-Plate Solar Collecto
EDGARALANA. DONASCO Solar Disinfection of

EDGAR ALAN A. DONASCO

Abstract

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Mechanical Engineering, from MSU-Marawi City, Philippines. L. DONASCO, Assistant Professor II, College of I

Introduction

T billion people in the developing countries do not have ready access to
adequate and safe water supply (WHO, 1998). Among these, 800
million live in the rural areas. Furthermore, included in the report is the three million annual deaths caused by diarrheal diseases, which are closely associated with lack of access to clean water, food and poor personal hygiene.

The booming bottled water business, as can be noticed by a lot of companies venturing into it, indicates public awareness and perception of the unreliability of community water supplies. Reported cases of drinking water contamination through ill-maintained water distribution system raised seri. ous public concern. Like in other developing countries, it shows the prevail. ing problem of the community water supply systems due to inappropriate choice of technology, unreliable supply and maintenance work, difficulties in procurement of fuel or spare parts and weak management structures (Wegelin 1994). Because of these reasons, most households resort to bottled water, boiling and use of expensive packaged water purifiers, while others simply take the risk from drinking tap water directly. Thus, household spending in ensuring water safety or for medical expenses is increased.

A suggested approach to this prevailing problem involves the increased use of appropriate low-cost disinfection technologies and installation of wa ter supply facilities whose operation and maintenance can be managed and sustained with local resources. It should be simple, effective, and inexpen sive and could be used by individuals, families or institutions like hospitals, offices and schools. A promising technology is the disinfection process through solar heating and radiation. The use of renewable energy such as solar energy is the basis of this low-cost and sustainable technology (Wegelin, 1994). It is therefore the intention of this study to investigate the possibility of fabricat ing locally a solar reactor and to test its performance.

Objective of the Research

The objective of this research is to study the local applicability of the solar disinfection process for drinking water using local materials.

The specific objectives are:

- 1. to design and fabricate a Solar Disinfection (SODIS) Plant for drink ing water application;
- 2. to determine the thermal performance of the SODIS Plant; and
- 3. to determine the effective operating parameters of the SODIS Plant for solar disinfection of drinking water that would pass WHO micro biological standards.

Scope and Limitations

This study on solar disinfection process for drinking water application is specifically intended for the Philippine setting and other tropical areas. The treatment process that the SODIS Reactor is designed to accomplish is sim ply microbiological disinfection; hence, only microbiological parameters and water quality standards were considered. An indicator organism was used to represent the state of microbiological quality of the raw and treated water of the solar disinfection process. The study did not track the fate of every mi croorganism present before and after the treatment process. Furthermore, the study focused on selected parameters such as temperalure, turbidity, expo sure duration, solar intensity, and Total coliform count as indicator of treat ment effectiveness.

Review of Related Literature

Solar Collector and Thermal Efficiency

A typical Flat-Plate Collector which consists of an optically transparent piate which acts as infrared trap, a space filled with air or vacuum, a heat absorber, and pipes through which air or fluid is circulated. Sometimes, in stead of pipes, a sheet of fluid is allowed to flow over the heat absorber plate (Schulert, 1981).

The major Constituent of any collector is the absorber. It is a metal, glass, or plastic surface where the absorbed solar radiation is converted to thermal energy and transferred by thermal conduction and convection to the coolant or circulating fluid.

To reduce the irradiative losses from an absorber while at the same time maintaining a high solar absorptance, a selective surface is often applied to the absorber. These surface coatings are composed of specially formulated

aints, chemical dips, or electroplated films that have the roperty of high absorptance (Neville, 1995). property of high absorptance (Neville, 1995). $\sim N_0$.

Microbiological Quality of Drinking Water
The diverse species of pathogenic agents are impossible to identify
enumerate individually. For this reason, indicator microorges:
to indicate their natural is indiffulnerate individually. For this reason, indicator microorganisms are used to indicate their potential presence.
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WHO (1993), in its Guidelines for Drinking Water Quality, standardized coli, WHO (1993), in its *Guidelines for Drinking Water Quality*, standardized E. coli, *Thermotolerant coliform* bacteria, and Total coliform bacteria $_{48 \text{ in}}$. dicators of bacteriological quality for drinking water.

WHO Microbiological Standard
The WHO standard for all water intended for drinking says that E , coli or Thermotolerant coliform bacteria should not be detectable in any 100 -ml sample. For treated water entering and within the distribution system, E . coli, sample. For treated water entering and within the distribution system, E coli, Thermotolerant coliform bacteria or Total coliform bacteria should not be detectable in any 100-ml sample (WHO, 1993).

Solar Disinfection Process
Solar exposure is an ancient disinfection practice used without profound understanding of the process. Solar disinfection is the process of inactivating and destroying pathogenic microorganisms present in water through solar exposure. Several experiments conducted reported vulnerability of microor ganisms to thermal heat (pasteurization) and radiation from the sun. Further more, thermal heat and radiation showed synergetic effect in the rate of re ducing microbial population present in water (Wegelin, 1994).

Thermal Effect

More recent studies of Sommer (1997) on Solar Pasteurization (SOPAS) shows that for effective disinfection of drinking water, it is necessary to heat it at 70°C and maintain it at that temperature for at least 15 minutes.

Radiation Effect

The inactivation of bacteria by UV radiation results primarily from the absorption of radiation by the deoxyribonucleic acid (DNA) of microorgan- $\lim_{n \to \infty}$ and subsequent demineralization of thymine bases in DNA. These thyming ne dimers distort the double helix conformation of DNA and may

replication, which effectively inactivates the bacteria

Because the only ultraviolet radiation efective in destroying bacteria is that which reaches the bacteria, the water should be relatively free from turbidity that would absorb the ultraviolet energy and shield the bacteria, To climinate falure of the radiation disinfection proccss due to turbidity factor, thin film approach is being used (Metcalf & Eddy, 1991).

The effective component of solar radiation involved in microbial destruc tion seems to be the near —ultraviolet (A) band $(320-400)$ nm) and to a lesser extent the visible band of violet and blue light (400-490 nm) (Acra, 1984).

Methodology

Equipment Set-up

The schematic diagram of the SODIS plant is shown in Figure 1. The plant location was at the rooftop of the Solar Laboratory of the University of

Figure 1. Schematic Diagram of SODIS Plant

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the Philippines campus in Diliman, Quezon City. f sample box w the Philippines campus eliminated inconvenience of sample handling and d_{e} .

SODIS Design and Fabrication

The primary considerations of the SODIS Plant design are transportabil. ity, thermal characteristic, operation and maintenance convenience, and as well as economic viability.

SODIS Reactor Details

The heart of the plant is the reactor (Fig. 2). It is where irradiation and heating of the water takes place. It is made of 3-mm thick aluminum plate. which has the advantage over ordinary steel plate in terms of susceptibility to corrosion and weight. It was painted with food-grade black paint (absorptivity = 0.9 to 0.98) on the surface in contact with the water subjected to treatment. Black has the optimum radiation absorptivity property. The di mensions of the single reactor used were 609 mm length x 1219 mm width for transportability and maintenance convenience. A food-grade 18 mm di ameter 0-ring, positioned on top of the aluminum plate perimeter, was used to both provide the volume space for water to be treated and at the same time to prevent water leakage. It is pressed by a heat-resistant glass (transmittance $= 0.79 - 0.85$) positioned on its top (see Figure below). Due to the space occupied by the O-ring, the remaining net reactor area is 0.675 m², or a volume capacity of 10 liters.

Instrumentation

The plant was provided with on-line sensors (Fig. 3), which were essent tial during experimentation. PT 100 thermocouple was used for water temperature monitoring, and a pyranometer (Type CM 11 by Kipp and Zonen, Holland with sensitivity of 5.55 x 10⁻⁶ V/(W/m²)) device for the solar intensity. The pyranometer was positioned with the same angular inclination oriity. The pyranometer was positioned with the same angularity. Both intation with that of the reactor. It gives output in terms of Watts/ m^2 . Both intation. instrumentation devices were properly calibrated prior to experimentation.

Thermal Performance Test of the Reactor

This test was incorporated in the microbiological test with considerations only on two physical parameters - water temperature and solar intentions only on two physical parameters - water temperature and solar intention June 2000

sity. With the aforementioned parameters, SODIS reactor heating efficiency and rate of temperature increase were determined.

Thermal Efficiency
Thermal efficiency gives the percentage of the amount of solar heat absorbed by water out of the total solar iradiation. It is given by the formula:

> Thermal Efficiency = $Q/SE * 100\%$; (1)

- 1. Angle Bar (50 mm x 25 mm x 3 mm thick)
- 2. Bolt (5 mm dia. X 30 mm L)
- 3. Heat Treated glass (4 mm thick)
- 4. Insulation (25 mm thick aluminized polyethylene)
- 5. Water Space (15 mm depth)
- 6. O-ring (18 mm dia.)
- 7. Bracket
- 8. Aluminum Absorber Plate (609mm x 1219mm x 4 mm thick)

Lower End

- 1. PT 100 sensor
- 2. Exit Hole (12 mm dia)
- 3. Inlet Hole (12 mm dia)
- 4. Rubber O-ring
- S. Air Ejector (4 mm dia copper tube)

 (2)

where:

Q: heat absorbed by water SE; solar energy exposure m; mass of water (11 kg) cp; specific heat constant of water $(4.187 \text{ KJ/(kg} \cdot ^{\circ}C))$ A ; area of reactor (m^2) Ti, Tf; initial and final temperature of water $(^{\circ}C)$ ASI; average solar intensity (KW/m^2) te, duration of solar exposure (second)

Rate of 1emperature Increase

ate of temperature increase is the amount of water temperature rise in the Celsius seale for a certain period of time with a determined amount of

solar energy exposure. It is determined by the formula below:

rate of temp. increase = $(Tf - Ti)/(ASI * t)$; °C/(KW-Hr/m²) (4)

> where: Tf: final temperature of water (C) Ti ; initial temperature of water $(^{\circ}C)$ ASI; average solar intensity (KW/m^2) 1: duration of solar exposure (hr)

Microbiological Test of Water Quality

Microbiological test was composed of two parts: Group A was aimed to find the minimum temperature setting for effective solar disinfection; and Group B was used to replicate the minimum temperature setting to further validate the findings.

Raw Water Preparation

For conservative experimental results, raw water used for the Group A microbiological test was a mixture of sewage and deepwell water. The sew age was practically intended to be the contaminant or the source of micro biological contamination, while the remaining portion, which is deepwell water, was intended to ensure that the raw water is unchlorinated, which could affcct the result of the experiment. Water in the UP Diliman Lagoon at the center of the UP Academic Oval lagoon comes from sewage of dormito ries and college buildings, which is contaminated with fecal matters. Propor tions of one-fourth Lagoon water and three-fourths deep well water was used to prepare the raw or untreated water for solar disinfection experiment. Deep well water was fetched every morning from Dagohoy Village, UP Diliman, Quezon City, which is about half kilometer away from the reactor site.

Raw water for Group B test, however, made use only of plain deepwell water - this time to fine tune the result to typical deepwell water, which is microbiologically contaminated.

Group A: Finding The Minimum Temperature Setting

cess. Batches were set at different levels of maximum temperature setting. Seven test runs were conducted within the 40 to 65°C range. Finding the minimum setting is essential in optimizing the operation of the solar disinfection pro

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Group B: Verification of the Minimum Temperature Setting

Inimum temperature setting result of $Group A$ was replacing result of $Group A$ 12) times in order to add confidence to the findings.

Temperature and Solar Intensity
Throughout the duration of water exposure to solar radiation, water temperature was logged at 10 minutes time interval together with the corresponding solar intensity.

Turbidity

Turbidity of the raw water was determincd prior to treatment, inas much as it affects radiation penetration to water depth. Turbidimeter test kit by Cole Parmer was used.

Microbiological Assessment

Raw and treated water samples were analyzed for their microbiologi cal content. Standard procedures for Multiple Tube Fermentation Technique (MTFT) which gives result in Most Probable Number (MPN), was used to determine the initial and final populations of Total coliform present in water before and after treatment (Philippine National Standards for Drinking Wa ter, 1993).

Sterilization

clave for 15 minutes maintained at 121°C. Sampling bottles, tubes and pipettes were sterilized using an auto-

Results and Discussion

SODIS Reactor Thermal Performance

reactor. The first set consisting of first seven test runs was considered. Eighteen test runs were conducted for the thermal ϵ _{per}formance of the test runs was σ conduction October 1999 while the rest of the test runs were done on June 2000.

Turbidity	Ave. Solar Intensity	Exposure Time	Final Temp	Initial Temp	Heat Absorbed	Thermal Effy.	Rate of Temp Rise
NTU	KW/m^2	Min	σ ^o σ	\overline{C}	KJ/kg	$\%$	$C/(kw/m^2)$ hr)
3.50	0.71	110	55.0	38.5	69	20	13
1.80	0.55	140	51.6	36.9	62	18	11
12.00	0.57	180	55.0	36.4	78	17	$\overline{11}$
34.00	0.49	90	45.2	34.7	44	22	4
9.00	0.67	190	60.0	33.8	110	19	12
9.00	0.66	130	55.0	33.8	89	23	15
25.00	0.42	180	50.2	34.3	67	20	13
20.00	0.46	180	46.0	35.5	44	12	8
12.00	0.67	50	50.6	41.1	40	26	17
13.00	0.54	100	50.2	36.8	56	23	15
12.00	0.45	100	48.1	36.7	48	24	15
9.00	0.47	140	50.6	31.0	82	28	18
10.00	0.42	160	50.0	35.9	59	20	13
12.00	0.55	110	53.1	36.0	72	27	17
13.00	0.82	60	52.0	37.0	63	29	18
11.00	0.71	70	50.4	38.1	52	23	15
8.00	0.54	90	50.9	36.1	62	29	18
9.00	0.80	70	51.1	38.0	55	22	4
12.41	0.58	119	51.39	36.14	64	22.15	14.28

Table 1. SODIS Thermal Performance

Highlighted are the average figures for each column

Figure 4. Reactor Thermal Efficiency

Thermal Eficiency

Example 1 Experiency of the SODIS Reactor as shown in Table 1 $falls$ with an average of 22.15 nergant $\frac{m_{\text{thm}}}{m_{\text{th}}}}$ the range of $17-23$ percent, with an average of 22.15 percent. This where based on 90% capacity of a 10-liter reactor amounting to an actual volume based on 90% capacity of the case of 0.675 m². Bansal (1990) generalized the 9 liters and a net surface area of low temperature solar collectors as between 20 to 30 percent. Thermal erroring of $17 - 23$ percent, with an average of 22.15 percent. This with

Thermal efficiencies in Table 1 are specific to the water temperature range of 32 - 60°C. Temperature of water heated in the reactor displays a relatively faster heat absorption rate at the start of the heating process, and hence. higher efficiency for that range. This is explained by the fact that reactor heat loss to the surrounding is proportional to its temperature gradient wüt the ambient. This temperature gradient (reactor mean temperature - ambient temperature) is minimum at the start of the heating process and increases as heating continues. Theoretically, continued heating would reach a maximum level (steady state condition) where heat gain is equal to heat loss (Schulert, 1881).

Turbidity variation of the treated water within the range of 1.5 NTU-34 NTU does not show significant effect on the reactor thermal efficiency.

Rate of Temperature Increase

Table 1 data too gives the computed rate of temperature increase of water inside the reactor; having a range of 8-18°C/(KW-Hr/m²) and an average of 14.28°C/(KW-Hr/m²) for an average solar intensity of 0.58 KW/m². It means that some 8.3°C temperature rise is expected per hour when solar intensity S about 0.58 KW/m². If raw water inside the tank is initially at 34° C, it would take about 2 hours to reach 50°C for an average solar intensity of .58 KW m

Microbiological Tests
Group A: Finding the Minimum Effective Temperature Setting Group A: Finding the Minimum Effective Temperature Setting
Table 2 shows the microbiological result of the solar disinfection treatment tests when raw water were subjected to different temperature levels for the nurmose of finding the subjected to different temperature levels for the purpose of finding the minimum temperature setting.

\boldsymbol{R} aw Water $\boldsymbol{\epsilon} = \boldsymbol{S} \boldsymbol{\beta} \boldsymbol{N}^{\boldsymbol{\beta}}$

Raw Water
All Total coliform tests on the raw water samples resulted to 5-5-5 per tive combinations for 10 ml, 1 ml and 0.1 ml sample dilutions in the

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confirmative phase of the MTFT. Such combination of positive tubes corre sponds to ³1600 MPN/100 ml. This is important to note in order to confirm the extent of contamination of the raw water.

Solar Exposure

Still refering to Tabie 2, treated water samples which yielded non-detect able Total coliform results indicate a minimum of 4600 KJ/m² of solar radiation dose (as white light). Failed sample results (10/13/99 and 10/15/99 data) had radiation doses lower than 4600 KJ/m². With a solar intensity of 0.71 KW/m', solar dose equivalent to 4600 KJ/m² is achieved in 110 minutes and 140 minutes if solar intensity is .55 KW/m'.

Turbidity

Turbidity of raw water samples ranged from 1.8 - 34 NTU with non distinguishable effect to the treatment effectiveness. Philippine National Stan dards for Drinking Water (1993) allows an acceptable turbidity level of 5 NTU, which is lower than the turbidity level of the majority of the raw water used. It is therefore expected that the solar reactor would even perform better microbiologically if raw water used has turbidity within the allowable stan dard. The raw water used was selected to exceed the normal turbidity level of drinking water so as to come up with a microbiological result that is on the conservative side.

Turbidity's non-distinguishable effect on the solar disinfection perfor mance is attributed to raw water depth setting inside the reactor, which is just 15 mm. Theoretically, turbidity filters off UV penetration to water's depth and in effect hampers the disinfection process. Making water depth consid erably shallow, however, compensates the negative effect of relatively high turbidity level of raw water. Data show that 15 mm water depth setting inside the solar reactor can accommodate turbidity variation as high as 38 NTU and still accomplish total inactivation of Total coliform.

Microbiological Assessment:

Table 2 further shows that 51.6°C is the minimum temperature to inacti vate Total coliform in the water sample. Test runs 10/13/99 and 10/15/99, with treatment temperatures at 48.6°C and 45.2°C, respectively, yielded com plete five tubes positive (316 MPN/100 ml) results in their treated water samples. Comparative result of experiment as reported by Sommer in 1995

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showed complete inactivation of Faecal coliform at 50°C. Wegelin (1994) reported significant synergetic effect on inactivation of bacteria at the same
temperature level.

Group B: Verification of The Minimum Tenperature Setting

Table 3 shows results of the additional thirteen (13) test runs conducted to further verify the minimum effective temperature setting found in Group A test where S1.6°C was the lowest. Compared to group A test, raw water used for this group test was plain deepwell water. In anticipation of a slighty contaminated state (no sewage component added), five (5) 10-ml sample dilution was used instead of the five sets for cach 10 ml, I ml, and 0.1 ml dilutions. All test runs

Raw Water (25%lagoon and 75%deepwell)				Treated Water				
Sampling Date	Temp. 0C	Turbid ity NTU	Total Coliform (MPN)	Temp. 0C	Total Coliform MPN	Exposure min	Solar Dose KJ/kg	Solar Intensity KWm ⁻
10/8/99	38.5	3.5	≥ 1600	55	≤ 2	110	4668	.71
10/9/99	36.9	1.8	≥ 1600	51	\leq 2	140	4636	55
10/10/99	36.4	12	≥ 1600	55	≤ 2	180	6115	57
10/13/99	32.5	3.5	21600	48.6	≥ 16	100	2924	.49
10/15/99	34.7	34	≥ 1600	45.2	≥ 16	90	2639	$\pmb{.6}^{\tau}$
10/26/99	33.8	$\overline{9}$	≥ 1600	55	≤ 2	190	7626	66
10/26/99	33.8	\mathbf{Q}	≥ 1600	60	\leq 2	130	5174	42

Table 2. Microbiological Test Group A: Finding the Minimum Effective Temp. Setting

Plain Deepwell Raw Water					Treated Water				
Sample	Turbidity	Temp.	MTF Result	Total Coliform	Temp.	MTF Result	Total Coliform	Exposure	Solar Dose
No.	NTU	\overline{C}	# of tubes positive	(MPN)	$^{\circ}$ C	# of tubes positive	MPN	min	KJ/kg
1	20	34.3	5	\geq 16	50.2	$\mathbf{0}$	≤ 2.2	180	4536
$\overline{\mathbf{c}}$	20	35.5	5	\geq 16	46.3	4	16	180	4968
3	16	31.4	5	\geq 16	50.7	\boldsymbol{l}	52.2	180	***
4	12	41.1	5	≥ 16	50.6	$\bf{0}$	≤ 2.2	50	2010
5	13	36.8	5	\geq 16	50.2	$\bf{0}$	≤ 2.2	100	3240
6	12	36.7	5	\geq 16	48.1	$\mathbf{0}$	≤ 2.2	100	2700
$\overline{7}$	9	31	5	\geq 16	50.6	$\mathbf{0}$	≤ 2.2	140	3948
8	10	35.9	5	\geq 16	50	$\mathbf{0}$	≤ 2.2	160	4032
9	12	36	5	\geq 16	51.3	$\bf{0}$	≤ 2.2	110	3597
10	13	37.0	5	≥ 16	52.0	$\mathbf{0}$	≤ 2.2	60	2938
11	11	38.1	5	\geq 16	50.4	$\boldsymbol{2}$	5.1	70	2982
12	8	36.1	5	\geq 16	50.9	$\bf{0}$	≤ 2.2	90	2889
13	9	38.0	5	\geq 16	51.1	$\bf{0}$	≤ 2.2	70	3352

Table 3. Microbiological Test Group B: Verification of the Minimum Temp. Setting

yielded all 5 tubes positive in the raw water confirmative test, which is equiva lent to \geq 16 MPN/100 ml. This confirms that deepwell water source at Dagohoy Village, UP-Diliman, is microbiologically contaminated.

result. Target temperature for Group B test was at least 50°C. However, test runs which somehow unable to reach 50° C due to weather condition were nevertheless examined. Among the 13 test runs, 10 yielded non-detectable treated water result for Total coliform (0 positive for 5 10-ml samples, equivalent to \leq 2.2 MPN/100 ml), one (sample no. 2) yielded 4 tubes positive (equivalent to 16 MPN/100 ml), one run (sample no.11) resulted to 2 tubes positive (equivalent to 5.1 MPN/100 ml), and another one (sample no.3) with I tube positive (equivalent to \leq 2.2 MPN/100 ml). Table 3, however, shows that sample which resulted to 16 MPN/100 ml (4 tubes positive) reached only a maximum temperature of 46.3°C. Considering only those samples that were able to reach 50°C, 10 out of 12 test runs or 83.33 percent yielded 0-positive

Turbidity

Turbidity of plain deepwell water used for Group B test ranged from $8 \text{ to } 20$ NTU. Consistent with Group A test, turbidity in that range did not show significant interference to the effectiveness of the disinfection process. That of course, is explained in Group A test, was due to the considerably thin depth of $_{\text{water}}$ $\frac{1}{2}$ used inside the reactor.

SODIS Plant Capacity
If calculations are based on the average solar data as reported by PAG-ASA for Quezon City Philippines for the year 1995 (average solar intensity of 0.70 KW/m² and an effective daily solar exposure of six hours), the solar reactors capacity is about 40 liters/m²/day or 13,049 liters/year liters drinking water per person per day consumption, a square meter of re actor could supply to about 27 persons.

Economic Implication

What makes solar disinfection technology attractive is the fact that solar energy is free and renewable. Using (LPG) Liquefied Petroleum Gas (HHV = 1000 BTU/ft³, boiling thermal efficiency of 70%)) to boil water, and with the current P265/11 kg LPG price, 1 liter boiled water would cost about P0.18. For each 1 $m²$ of reactor area, which has about 40 liters a day output capacity, a household is able to save about P7.00 per day. If solar reactor is allowed to operate at 0.90 utilization factor, savings from boiling with LPG is about P2400.00 per year. Such an amount is equivalent to nine (9) 11-kg LPG tanks savings per year.

Savings could more be emphasized if comparison is done with bottled water which costs about P75.00/20 liters (Manila Price, September 2000; 1US\$=P45), or an equivalent savings of P150.00 per day. For one year solar reactor production output, equivalent bottled wáter cost is P97,867.42.

Environmental implication of using solar disinfection for drinking water includes mainly the reduction on the utilization of non-renewable SOurces energy, and the avoidance of combustion products, which contribute to the deterioration of the environment. Calculations show that 98 m^3 of CO_2 could be formed annually if disinfection through boiling is done by m means of LPG

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Fabrication Cost Estimate and Payback Period

Labor and materials costs incurred in the fabrication of the SODIS Plant amounts to P6,190.00. If payback period is computed based on the cost of boiling water through LPG, capital investment is realized within 3.81 years.

Conclusion and Recommendations

The results of the study show that local application of solar disinfec tion technology is a viable alternative for microbiological treatment of drinking water. The design and fabrication of the reactor, which is a flat-plate solar collector, is relatively simple. It is designed with features måking it easy to be transported, ease of operation and maintenance, and of reasonable fabri cation cost.

Specific conclusions based on the results of the study are:

- 1. The SODIS reactor operates at a satisfactory average thermal effi ciency of 22 percent.
- 2. Rate of temperature increase is 10°C per hour based on average solar intensity data of Quezon City, Philippines, which is 0.70 KW/m².
- 3. Effective solar disinfection of drinking water is achieved, passing WHO Microbiological Standard, which is for Total coliform to be non-detectable when water temperature is set to at least 50°C. At 0.70 KW/m' solar intensity, exposure duration is about 2 hours.
- 4. Turbidity as high as 38 NTU did not affect the results with water depth setting inside the reactor which was 15 mm.
- 5. Shifting from boiling drinking water using LPG into solar disinfec tion can offer around 4 years payback on the fabrication cost of the SODIS Plant.

Considering the time and financial resource limitations during the course of study, some advantageous features were not tried. Hence, there is still much room for design enhancement for the purpose of performance optimi zation without compromising the simplicity of the process. The following recommendations are proposed to improve the solar reactor performance:

1. Use of solar reflectors for redirection of solar incident rays to in

crease solar rays concentration heating the reactor, and
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- 2. Development of a device to address the fouling of the reactor $glas$ cover, which can reduce the transmissivity of radiation.
- 3. Undertaking the study in other areas of the country to further value date the results incorporating thermal design enhancements.

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